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### **Thermodynamics worked examples Taylor Francis**

Thermodynamics worked examples 1. What is the absolute pressure, in SI units, of a fluid at a gauge pressure of 1.5 bar if atmospheric pressure is 1.01 bar? Solution Absolute pressure =  $p = p_g + p_a = 1.50 + 1.01 = 2.51 \text{ bar} = 251 \text{ kPa}$  2. Convert -25 oC to a temperature in degrees Kelvin. Solution  $T[\text{K}] = T[\text{oC}] + 273.15 = -25 + 273.15 = 248.15\text{K}$  3. Calculate the following: (i) the kinetic energy of

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27 | Biochemical Thermodynamics The branch of physical chemistry known as thermo-dynamics is concerned with the study of the trans-formations of energy.

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The book includes all the subject matter covered in a typical undergraduate course in engineering thermodynamics. It includes a series of worked examples in each chapter, carefully chosen to expose students to diverse applications of engineering thermodynamics. Each worked example is designed to be representative of a class of physical problems. At the end of each chapter, there are an additional 10 to 15 problems for which numerical answers are provided.

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M. Bahrami ENSC 388 (F 09) Intro and Basic Concepts 1 Basic Concepts of Thermodynamics Every science has its own unique vocabulary associated with it.

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